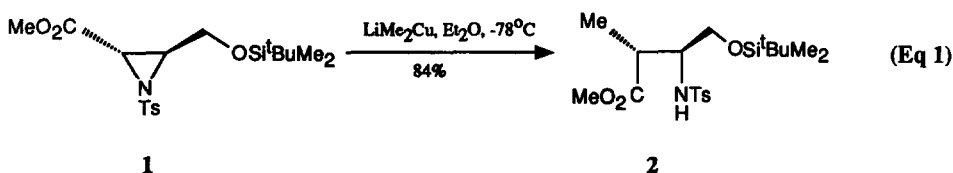


NUCLEOPHILIC RING OPENING OF C_2 -SYMMETRIC AZIRIDINES.
SYNTHETIC EQUIVALENTS FOR THE β -CATION OF ASPARTIC ACID

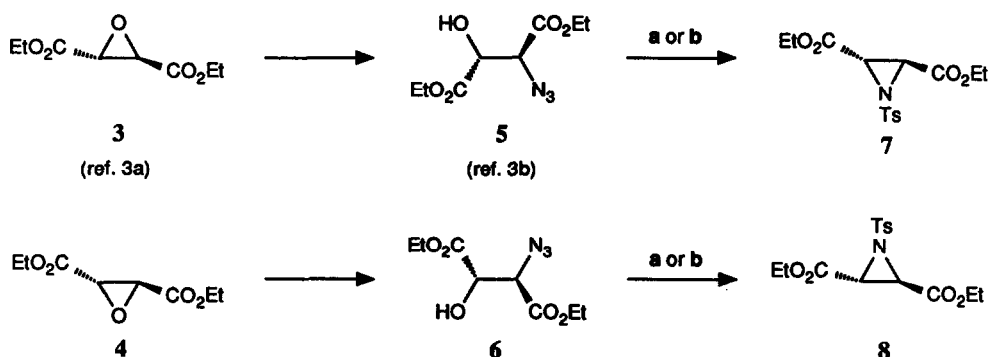
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Abstract: The non-racemic C_2 -symmetric aziridines **7** and **8**, easily available from (+)- and (-)-tartaric acid, respectively, undergo rapid nucleophilic attack to yield products formally derived from the β -cation of L- or D-aspartic acid.

The synthesis of novel amino acids is currently a topic of great interest^{1a-c} and both we^{2a,b} and others^{2c-g} have demonstrated the utility of suitably functionalized chiral aziridines as key synthetic intermediates. In the course of our work on the enantioselective synthesis of β -lactam antibiotics^{2b} we recently attained complete regiocontrol in the ring opening of the aziridine shown in Eq 1.



The excellent regioselectivity shown above prompted us to synthesize the C_2 -symmetric aziridines **7** and **8** (Scheme) for reaction with nucleophiles, in order to obtain products formally derived from the β -cation of either L- or D-aspartic acid.

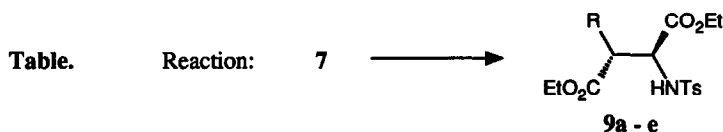


Scheme. (a): (i) PPh_3 , C_6H_6 , reflux \rightarrow N-H aziridine (ii) $pTsCl$, pyridine, 73% overall.
(b): (i) PPh_3 , C_6H_6 , RT \rightarrow amino-alcohol (ii) $pTsCl$, pyridine, $-20^\circ C$ \rightarrow N-tosyl amino-alcohol
(iii) DEAD, PPh_3 , THF, RT, 61% overall.

In this Letter we present some of our results which complement those published in two recent papers by Baldwin^{1c,2c}. In the first of these^{1c} Baldwin showed that the β -anion of aspartic acid provides a useful entry to non-proteinogenic amino acids, while in the second paper^{2c} he describes the ring opening of aziridine-2-carboxylate esters with organometallic reagents.

Our route to the chiral aziridines **7** and **8** is shown in the Scheme, key intermediates being the known^{3a} epoxides **3** and **4** and the azido-alcohols^{3b} **5** and **6**. The latter were transformed to the desired aziridines^{3c} in two different ways, one of which involved an efficient intramolecular Mitsunobu reaction⁴ to close the three-membered ring.

The C₂-symmetry of **7** and **8** removes the issue of regiochemistry from any discussion of ring opening, and nucleophilic attack on these species was expected to occur with clean inversion², thus conserving stereochemical and enantiomeric purity. Finally, the two ester groups and the tosyl moiety were expected to powerfully activate these aziridines towards nucleophiles. Our results are shown in the Table.



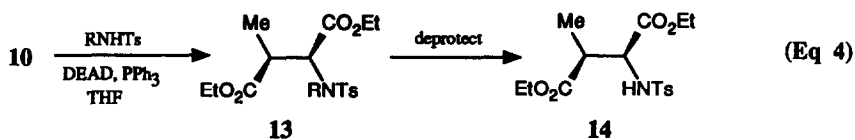
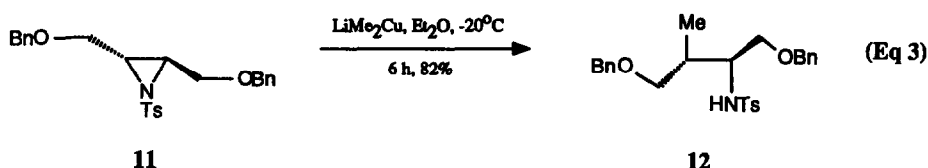
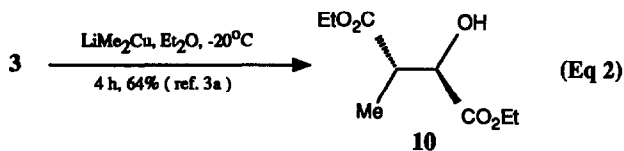
Entry	Nucleophile	Solvent	Temp.	Product	R	Isolated yield ^a
1	LiMe ₂ Cu (2 eq.)	Et ₂ O	-78°C	9a ^{5a}	Me	68%
2	LiBu ₂ Cu (2 eq.)	Et ₂ O	-78°C	9b	Bu	54%
3	Li ₂ Bu ₂ CuCN	Et ₂ O	-78°C			^b
4	NaN ₃ (2 eq.)	DMF	30°C	9c ^{5b}	N ₃	81%
5	MgI ₂ (3 eq.)	THF	0°C	9d	I	72%
6	MgBr ₂ (3 eq.)	THF	0°C	9e	Br	76%

^a After flash chromatography. ^b Decomposition.

The first nucleophiles tested (entries 1 and 2) were the Gilman cuprates, and it was quickly found that **7** and LiMe₂Cu reacted extremely rapidly (< 2 min) at -78°C in ethereal solution to furnish **9a** in acceptable isolated yield. The dibutyl cuprate gave somewhat poorer yields, partly due to the concomitant formation of the reduction product N-tosyl diethylaspartate⁶. Use of the Lipshutz higher-order cyanocuprates⁷ (entry 3) was completely unrewarding. The facility with which **7** underwent ring-opening by the Gilman cuprates is remarkable, and stands in sharp contrast to the relatively sluggish behaviour of both its epoxide counterpart **3** (Eq 2) and the aziridine⁸ congener **11** shown in Eq 3. For further comparison it may be noted that the reaction shown in Eq 1 was complete after 3h at -78°C. The combined activating power of the ester groups and the N-tosyl moiety of **7** is thus clearly demonstrated.

A further contrast between aziridine **7** and epoxide **3** is provided by entry 4 in the Table, since **3** is essentially inert towards sodium azide under these reaction conditions (see also Ref. 3b). Iodide and

bromide (entries 5 and 6) also ring-opened **7** under mild conditions and in high yield. The relevant *enantiomeric* series of products is provided by analogous reactions of **8** (the optical antipode of **7**) and we are currently attempting to obtain *epimeric* materials by the type of sequence⁴ shown in Eq 4.



In summary, we have demonstrated that the enantiomeric pair of C_2 -symmetric¹⁰ aziridines **7** and **8**, easily available in quantity from (+)- and (-)-tartaric acid, respectively, represent convenient synthetic equivalents for the β -cation of either *L*- or *D*-aspartic acid. Reactions of these aziridines with a wider range of nucleophiles are now being studied.

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4. For recent related examples, see: Henry, J.R., Marcin, L.R., McIntosh, M.C., Scola, P.M., Harris, G.D., Jr. and Weinreb, S.M. *Tetrahedron Lett.* **1989**, *30*, 5709.
5. (a) Selected data for **9a**: $[\alpha]_D +18.5^\circ$ ($c=0.53$, CH_2Cl_2); $^1\text{H NMR}$ (CDCl_3 , TMS): δ 1.04 (3H,t, $J=7$), 1.22 (3H,t, $J=7$), 1.26 (3H,d, $J=7$), 2.40 (3H,s), 3.11 (1H,qd, $J=7,4$), 3.94 (2H,q, $J=7$), 4.05 (1H,dd, $J=9.5,4$), 4.09 (2H,m), 5.49 (1H,d, $J=9.5$), 7.27 and 7.75 (2x2H, AA'BB', $J_{AB}=8.8$). IR: 3285(b), 1736(s), 1341(s), 1165(s) cm^{-1} ; (b) Selected data for **9c**: $[\alpha]_D +78.9^\circ$ ($c=0.8$, CH_2Cl_2); $^1\text{H NMR}$ (CDCl_3 , TMS): δ 1.11 (3H,t, $J=7$), 1.30 (3H,t, $J=7$), 2.40 (3H,s), 4.05 (2H,m), 4.26 (2H,q, $J=7$), 4.43 (1H,d, $J=3$), 4.46 (1H,dd, $J=7.5,3$), 5.49 (1H,d, $J=7.5$), 7.30 and 7.76 (2x2H, AA'BB', $J_{AB}=8.9$); IR: 3328(b), 2121(s), 1747(s), 1348(s), 1166(s) cm^{-1} .
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8. Aziridine **11** ($[\alpha]_D +1.57^\circ$, $c=0.98$, CH_2Cl_2) was prepared from the relevant non-racemic epoxide⁹ via the procedure outlined in refs. 2a and 2b.
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